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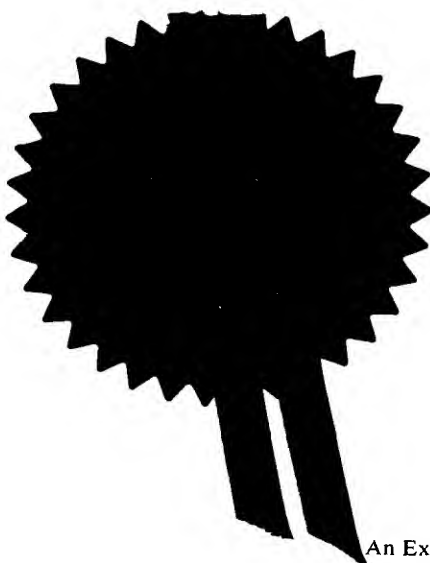
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29 APR 1999

30APR99 E443650-4 D01559
F01/7700 0.00 - 9909853.5

Request for grant of a patent

(See the notes on the back of this form. You can also get an explanatory leaflet from the Patent Office to help you fill in this form)

The Patent Office

Cardiff Road
Newport
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1. Your reference

MJ/TM/STS.32

2. Patent application number

(The Patent Office will fill in this part)

9909853.5

3. Full name, address and postcode of the or of each applicant (underline all surnames)

Surface Technology Systems Limited,
Imperial Park,
Newport,
NP1 9UJ.

Patents ADP number (if you know it)

07539679001

If the applicant is a corporate body, give the country/state of its incorporation

United Kingdom

4. Title of the invention

"Gas Generation System"

5. Name of your agent (if you have one)

Wynne-Jones, Laine & James

"Address for service" in the United Kingdom to which all correspondence should be sent (including the postcode)

22, Rodney Road,
Cheltenham,
Gloucestershire.
GL50 1JJ

Patents ADP number (if you know it)

1792001

6. If you are declaring priority from one or more earlier patent applications, give the country and the date of filing of the or of each of these earlier applications and (if you know it) the or each application number

Country

Priority application number
(if you know it)

Date of filing
(day / month / year)

n/a

7. If this application is divided or otherwise derived from an earlier UK application, give the number and the filing date of the earlier application

Number of earlier application

Date of filing
(day / month / year)

n/a

8. Is a statement of inventorship and of right to grant of a patent required in support of this request? (Answer 'Yes' if:

YES

a) any applicant named in part 3 is not an inventor, or
b) there is an inventor who is not named as an applicant, or

c) any named applicant is a corporate body.

See note (d))

Patents Form 1/77

9. Enter the number of sheets for any of the following items you are filing with this form. Do not count copies of the same document

Continuation sheets of this form	-
Description	10
Claim(s)	1
Abstract	-
Drawing(s)	2 + 2

10. If you are also filing any of the following, state how many against each item.

Priority documents	-
Translations of priority documents	-
Statement of inventorship and right to grant of a patent (Patents Form 7/77)	-
Request for preliminary examination and search (Patents Form 9/77)	1
Request for substantive examination (Patents Form 10/77)	-
Any other documents (please specify)	-

11. I/We request the grant of a patent on the basis of this application.

Signature	<i>Wynne-Jones, Laine & James</i>	Date	28.4.1999
Wynne-Jones, Laine & James			

12. Name and daytime telephone number of person to contact in the United Kingdom

Mr. M.J.G. James
01242-515807

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"Gas Generation System"

The industry demands for alternative process gases for dry (plasma and non-plasma) process systems has led to the investigation of a number of different chemistries. Several different highly-reactive halogen-based chemistries are known to be likely candidates to achieve an improved process capability including etch rate, selectivity etc. Gases in this category include pure-halide compounds and suffer from increased cost (of the delivery and abatement system as well as precursor itself), greater health and safety risks, transportation difficulty (and cost) as well as relatively poor commercial availability.

According to the invention there is provided a gas generator system comprising supply sources for halogen gases, including pure halide compounds, connected into a gas reaction chamber or chamber system enabling generation of a predetermined gas for use in a subsequent process, the reaction chamber having a valved outlet for direct supply of the generated gas to a single or multiple processing tool or process chamber.

More than one generator may be used for each process chamber.

In one embodiment, this invention will provide for the localised generation of reactive process gases. These can be generated local to each process tool through the direct combination of different precursor gases under controlled temperature and pressure reaction conditions. Various pure-

halogen and compound halide gases can be generated in this way. This system can include a feedback system to precisely control the precursor that is being generated.

A second embodiment of this invention will provide for
5 the local on-board synthesis of reactive precursors for plasma and non-plasma processing which include precursors which would otherwise not be readily available due to instability or short-lifetime or difficulty in commercial manufacture or supply. This allows for the direct synthesis
10 and reactive investigation of new groups of halide precursors for semiconductor or non-semiconductor processing.

A third embodiment of this invention provides for the synthesis of precursors that replace directly or indirectly (by simulating the behaviour) the gases that are either in
15 use or have been used in industry and are subject to restriction in use or withdrawal by new environmental legislation. The Montreal Protocol defines environmentally unfriendly chemicals including those that have already been restricted or those that will be restricted in application
20 in the near future. This will impact on the availability of the gases as production is restricted and also on the increasing necessity for abatement during even limited use. Indeed, much recent work has focused on the search for suitable alternative replacement gases (Fracassi and
25 d'Agostino, J.Vac.Sci.Technol.B 16(4), Jul/Aug. 1998). With suitable abatement means, the synthesis of the material on demand may well be an acceptable means of continuing to use existing chemistries. Commercially this may be the only

solution available.

The direct combination gas generator can produce a gas in the process chamber by reacting suitable precursor gases possibly with other materials that may be preloaded within
5 a temperature and pressure controlled reaction chamber that is local to a processing tool or chamber. The precursor gases may not react with the substrate being processed, or indeed be desired in the processing tool individually. The design of the system avoids the occurrence in the processing
10 tool or chamber of any possible adverse reactions, during the combination of the precursor gases, that might otherwise prejudice the overall process. Bypass of the process chamber to an abatement tool inhibits the pre-reaction gas mixture from adversely influencing the process in the
15 process chamber. The precursor gases are redirected into the process chamber by the process controller after the required flow and composition has stabilised.

The precursors for the reaction chamber include any of the following gases:-

- 20
1. at least one halogen selected from F_2 , Cl_2 , Br_2 , I_2 as the required halogen source;
 2. additional gases from N_2 , H_2 , O_2 ;
 3. additional gases from He, Ar, Kr, Xe.

The generator may further comprise any one or more of
25 the following components:-

1. electrode materials selected from C, B, S, Si, Ge, P in either elemental or appropriate compound form. Where appropriate, the electrode material

may be porous to increase the surface reaction area. The electrodes may be independently temperature controlled and electrically and/or magnetically biased;

- 5 2. appropriate catalysts (product generation dependent);
3. gas analysers such as infra-red spectrometers and/or mass spectrometers;
4. a feedback control system which senses the output
10 gas and adjusts reaction parameters (gas flow rates, temperatures, pressure etc) to achieve the desired output concentration;
5. an appropriate reaction vessel or chamber. A number of chambers may be used in series and/or
15 parallel to generate precursors requiring more complex generation mechanisms;
6. an appropriate vacuum control system, valving and gauging, including means for bypassing and purging;
- 20 7. an appropriate abatement system.

The generator can operate within the following temperature and pressure range: room temperature to several hundred degrees C at several Torr to atmospheric pressure. In particular the reaction chamber may be operated at or near
25 atmospheric pressure, going up the range from several Torr to 760 Torr. The reaction chamber temperature can be controlled at between ambient room temperature up to 600°C generally, but probably will lie within the range of 100 -

400°C. Differing temperatures may be maintained in at least 2 separate zones of the reaction chamber. For certain processes, elements of the system may be below room temperature for product purity control. The temperature control of any electrode may be separate from the vessel and the electrode may be independently biased either electrically and/or magnetically. This or other biasing means may be used to generate a plasma.

The supply to the gas reaction generator of high purity gases removes the need to polish the generated gas to remove unwanted impurities before passing into the process chamber. Mass flow controllers may be used to precisely meter the flow of the supply gases.

The valved control of the flow into the process chamber (if necessary augmented by a control system) ensures that the generator does not experience the low pressure (vacuum) at the process chamber. The preferred method of production of fluorine is by fluoride molten electrolytic generators which can be installed locally to the tool and overcomes the need to obtain high purity 100% fluorine in cylinders. Other means of F_2 generation can be appropriately used. The preferred method of supply of Cl_2 is from either an electrolytic generator or from high-pressure cylinders, both of which are readily available. Similarly, appropriate sources of Br_2 , I_2 may be used.

Multiple or singular gas inlets and/or either two or more electrodes (or a single composite electrode) as necessary may be used to generate some of the more complex

halide species. Electrodes within the reaction chamber may be electrically and/or magnetically biased.

The reaction chamber can be formed from high purity materials (such as those sold under the Trade Marks Monel (nickel/copper/iron alloy), Inconel (nickel/chromium/iron alloy) and Hastalloy (nickel/molybdenum/chromium/manganese /iron alloy)) which would not be financially feasible with large scale generation systems

The concept of local generation allows the production of the desired process gases to a high purity, although the specific reaction products may be a family of compounds not just one specific product. This can be achieved at a reasonable cost and low risk. The addition of this gas generator feeding a process tool, allows a novel capability to introduce different generated process gases into the process chamber where, for many different reasons, the option of using the gas directly would not be possible.

The gas generator is able to provide a local supply of various gases on demand, such as :-

1. Pure halogen compounds including Cl_xF_y , F_xBr_y , Cl_xBr_y and $\text{F}_x\text{Cl}_y\text{Br}_z$ (where x , y and z are integers). An I_2 source could also be added to provide additional I-containing compound halogens;
2. Groups of halide precursors for plasma and non-plasma processing which include precursors which would otherwise be unavailable due to instability or short-lifetime or difficulty in commercial manufacture or supply;

3. Freons or halo-carbides including C_xF_y , $C_wF_xCl_y$, $C_wF_xBr_y$, $C_wF_xCl_yBr_z$, C_wCl_x , $C_wCl_xBr_y$, $C_wF_xBr_y$, C_wBr_y (where w , x , y and z are integers). An I_2 source could also be added to provide additional I-containing halo-carbides;
4. Other reactive halides where the C is replaced by N or B;
5. Other reactive halides where any C is replaced by combination of C with N, B and/or H;
6. Metal halides, where the C source is replaced by an appropriate metal electrode;
7. Halide complexes of any combination of the groups listed in 3,4,5 and 6 above.

Ideal gases for generation include ClF , ClF_3 , ClF_5 , BrF_3 , BrF_5 , CF_4 , C_2F_6 , C_3F_8 , CHF_3 , CF_2H_2 , CCl_2F_2 , CCl_4 , CF_3Br and NF_3 , and related compounds.

The generator of the invention offers many advantages over direct gas delivery. For some of the gases the reaction gas generator can operate at or near atmospheric pressure, thereby eliminating the need for high pressure regulators on the system. Installation and storage of hazardous gases is reduced. The local generation, on demand, of the hazardous gases eliminates long (and usually high-pressure) gas lines from a central store to the installation and minimises associated safety risks. There would be a significant reduction in the installation costs due to the reduced level of plumbing of additional gases and the associated safety requirements. The maintenance of the

complete system is eased by the absence of any of the generated gases when the system is not operating. The quantity of the generated gases can be regulated to that required for the specific application so that the gas consumption is optimised and excess generated gas avoided. Some gases may require higher generation pressures (above atmospheric) and hence necessitate the use of pressure regulatory devices.

The invention may be performed in various ways and preferred embodiments thereof will now be described, by way of example, with reference to the accompanying drawings, in which:-

Figure 1 is a diagrammatic illustration of a general gas generation system of this invention;

Figure 2 indicates various input and control features of the system of Figure 1;

Figure 3 shows a modified system for that indicated in Figure 2; and

Figures 4, 5 and 6 illustrate still further modifications of the general gas generation system illustrated in Figure 1.

In the arrangement shown in Figure 1 precursor gases are passed from suitable supply sources 1 and 2 to a reaction chamber 3 where the gases are combined under controlled conditions. The reaction product is then fed to a process chamber 4 where a dry process utilising that gas is to take place. Appropriate valving will include valves provided at A, B, C and D for appropriate control and

isolation means. Linked control systems 5 and 6 monitor and maintain the supply to and the conditions in the chambers 3 and 4.

From the process chamber 4 gases pass to an exhaust system 7, which in turn leads to an abatement tool 8 (which is usually needed). A bypass outlet 9 leads from the reaction chamber 3 to the exhaust system, whereby gases can be switched into the process chamber only when required for processing. This also allows the means for ensuring stable gas composition and flow to be maintained prior to switching into the process chamber.

Figure 2 shows features of the reaction chamber 3, and connections thereto, in greater detail. This illustrates inlets 1A and 2A from sources 1 and 2 and another inlet 10 for a third precursor gas (there could be still further inlets if required). A control system 11 (which is a single system but for clarity is shown in two parts 11 and 15) can take measurements from analysis devices 12 and 13 and a pressure measuring device 14. The control system 15 can take measurements from a temperature measurement device 16 and also provide control for an auto pressure control device 17 and a temperature control device 18. Optionally an electrode 19 can be provided within the reaction chamber 3 with biasing means (not shown) and a temperature control device (not shown).

Figures 3, 4, 5 and 6 illustrate various ways in which electrodes can be incorporated into the reaction chamber 3 (which in turn may comprise a single chamber or a combina-

tion of chambers linked in serial or parallel or any desired combination thereof). In Figure 3 the chamber 3 is shown as being provided with a porous electrode 20. Figure 4 illustrates a two chamber system wherein the preliminary reaction chamber 3A incorporates a normal electrode 21 whilst a downstream chamber 3B incorporates a porous electrode 22. The third gas supply line 10 is connected to this secondary chamber 3B. In Figure 5 three linked reaction chambers 3A, 3B and 3C are shown, each incorporating a standard electrode 23, 24 and 25 with respective gas supply sources 3, 4 and 10. In the modified arrangement shown in Figure 6 the chambers 3A, 3B and 3C are modified to incorporate porous electrodes 23A, 24A and 25A.

CLAIMS

1. A gas generator system comprising supply sources for halogen gases, including one or more pure halide compounds, connected into a gas reaction chamber or chamber system enabling generation of a predetermined gas for use in a subsequent process, the reaction chamber having a valved outlet for direct supply of the generated gas to a single or multiple processing tool or process chamber.
2. A gas generator system as claimed in claim 1, which provides for the localised generation of reactive process gases.
3. A gas generator system as claimed in claim 1, which provides for the local synthesis of reactive precursors for plasma and non-plasma processing which include precursors which would otherwise not readily available due to instability or short-lifetime or difficulty in commercial manufacture or supply.
4. A gas generator system as claimed in claim 1, which provides for the synthesis of precursors that replace directly or indirectly (by simulating the behaviour) the gases that are either in use or have been used in industry and are subject to restriction in use or withdrawal by environmental legislation.
5. A method of generating gases by utilising the system as defined in any one of claims 1 to 4.
6. Any novel combination of a gas generator or a process for generating gases and substantially as herein described with reference to the accompanying drawings.

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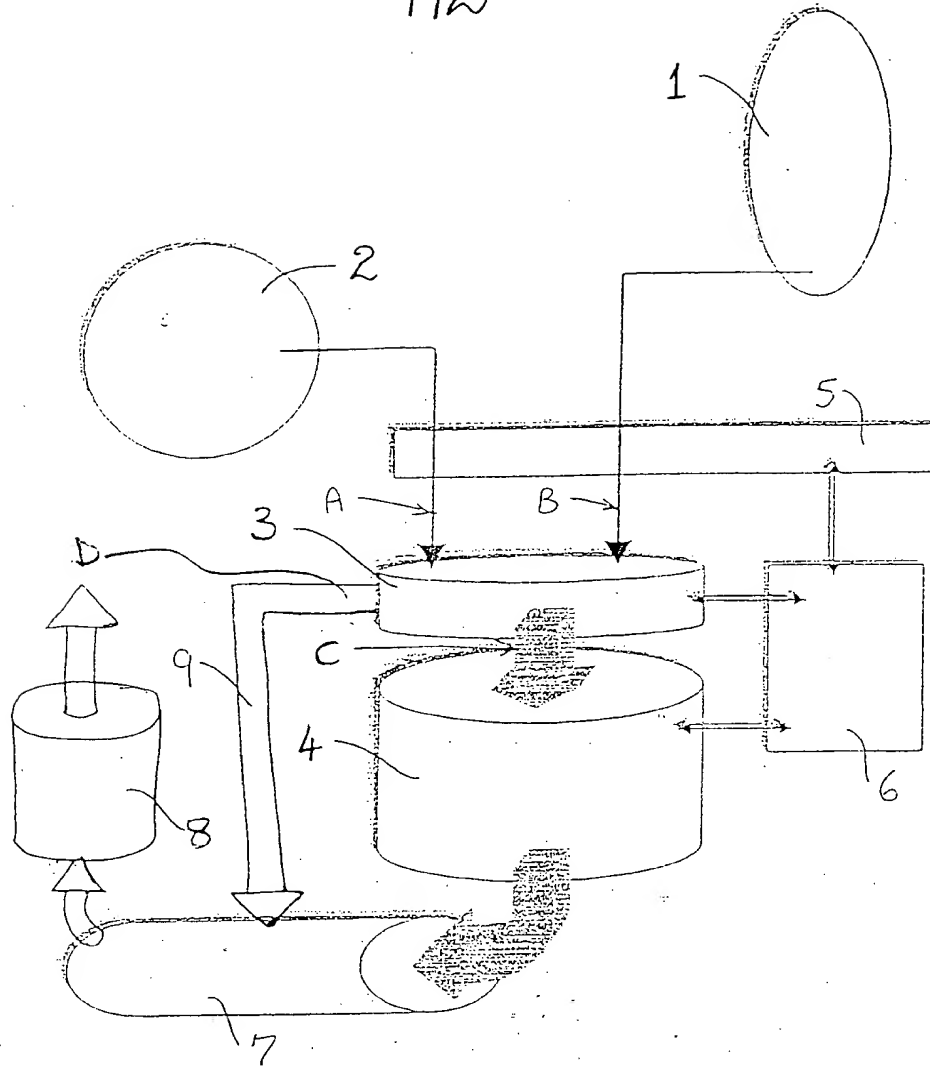


FIG 1

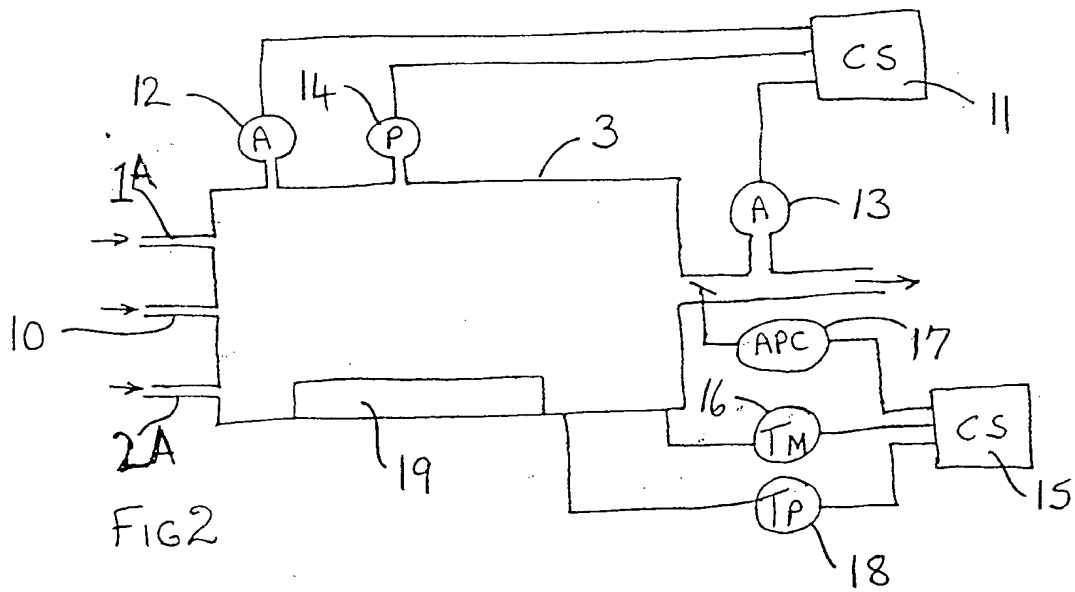


FIG 2

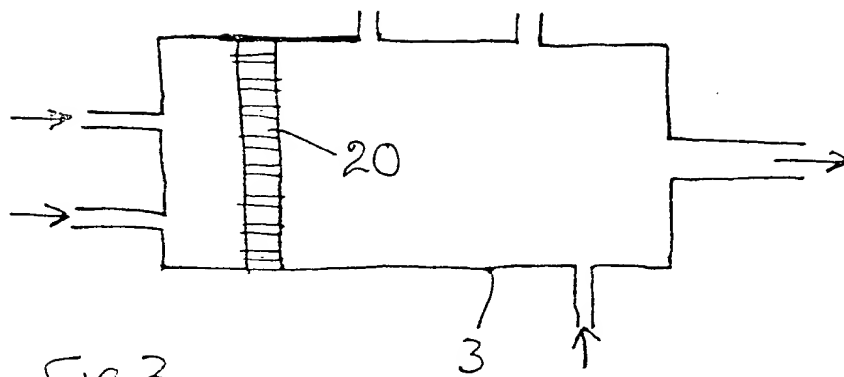


FIG. 3

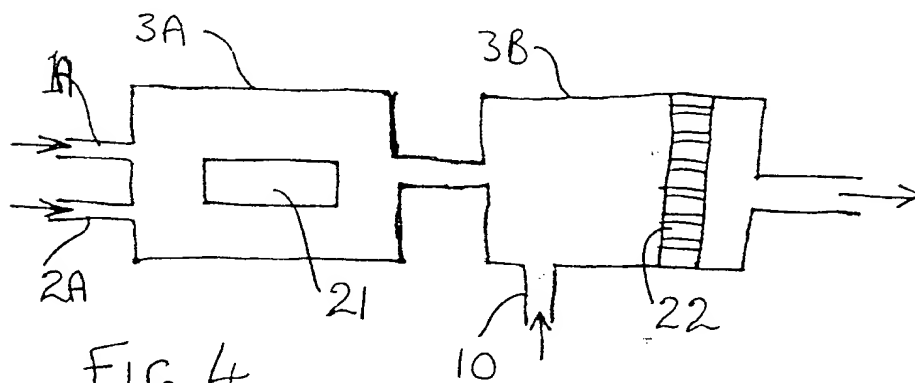


FIG. 4

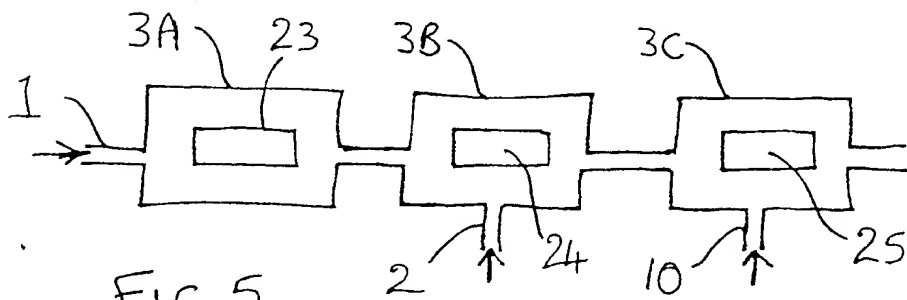


FIG. 5

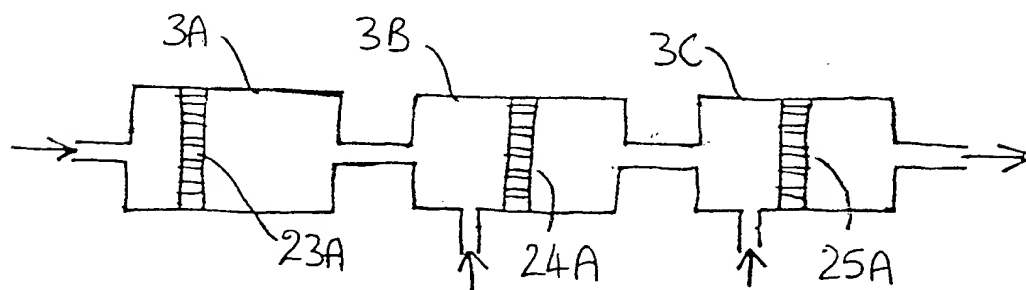


FIG. 6

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